

## CHEM6002: Chemical Principles

## Module Details

Module Code:	CHEM6002
Title:	Chemical Principles <b>APPROVED</b>
Long Title:	Chemical Principles
NFQ Level:	Fundamental
Valid From:	Semester 1 - 2019/20 ( September 2019 )
Duration:	1 Semester
Credits:	5
Field of Study:	4421 - Chemistry
Module Delivered in:	<a href="#">11 programme(s)</a>
Module Description:	This module introduces students to the fundamentals of atomic theory, chemical bonding, the periodic table, physical states of matter, and stoichiometric calculations

## Learning Outcomes

On successful completion of this module the learner will be able to:

#	Learning Outcome Description
LO1	Describe fundamental principles of atomic theory and chemical bonding
LO2	Outline the main features and trends in the periodic table of elements
LO3	Apply fundamental chemical concepts to writing chemical equations and naming inorganic compounds
LO4	Perform basic stoichiometric calculations
LO5	Use a range of chemical equipment and techniques to perform basic laboratory procedures

## Dependencies

## Module Recommendations

13215	CHEM6002	Chemical Principles
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## Incompatible Modules

No incompatible modules listed

## Co-requisite Modules

No Co-requisite modules listed

## Requirements

No requirements listed

## Indicative Content

## Atomic Structure

Sub-atomic particles, electronic structure of atoms, isotopes, radioactivity

## Periodic Table

Structure of the periodic table, classification of the elements, trends in the properties of the elements

## Chemical Bonding

Introduction to valence bond theory; ionic and covalent bonds, hybridisation, molecular geometry, valence shell electron pair repulsion theory (VSEPR); overview of metallic bonds and bonding in semi-conductors; relevant examples

## States of Matter

Intermolecular forces; physical properties of gases, liquids and solids

## Chemical Nomenclature

Aims, history and types of chemical nomenclature. Organic and Inorganic nomenclature.

## Stoichiometry

Balanced equations, the mole, solutions, concentration calculations

## Practical programme

General chemistry laboratory procedures, gravimetric and volumetric analysis

## Module Content &amp; Assessment

Assessment Breakdown	%
Coursework	100.00%

## Assessments

Coursework			
Assessment Type	Multiple Choice Questions	% of Total Mark	20
Timing	Week 7	Learning Outcomes	1,2,4
Assessment Description Theory assessment. Typical content examined: atomic structure, periodic table, chemical bonding states of matter, stoichiometry			
Assessment Type	Short Answer Questions	% of Total Mark	30
Timing	Week 12	Learning Outcomes	1,2,3,4
Assessment Description Theory assessment. Typical content examined: atomic structure, periodic table, chemical bonding, states of matter, chemical nomenclature, stoichiometry			
Assessment Type	Practical/Skills Evaluation	% of Total Mark	40
Timing	Every Week	Learning Outcomes	4,5
Assessment Description Performance of practicals and submission of reports and calculations			
Assessment Type	Practical/Skills Evaluation	% of Total Mark	10
Timing	Week 13	Learning Outcomes	4,5
Assessment Description Lab Exam to include independent performance of an experiment carried out in the practical programme and completion of associated calculations			
No End of Module Formal Examination			

<b>Reassessment Requirement</b>
<b>Repeat examination</b> Reassessment of this module will consist of a repeat examination. It is possible that there will also be a requirement to be reassessed in a coursework element.

## Module Workload

Workload: Full Time					
Workload Type	Contact Type	Workload Description	Frequency	Average Weekly Learner Workload	Hours
Lecture	Contact	Delivery of content and material underpinning learning outcomes	Every Week	3.00	3
Lab	Contact	Practical skills development	Every Week	2.00	2
Independent & Directed Learning (Non-contact)	Non Contact	Review of course material, completion of laboratory reports and exam practice	Every Week	2.00	2
Independent & Directed Learning (Non-contact)	Non Contact	Review of course material, completion of laboratory reports and exam practice	Every Week	2.00	2
Total Hours					9.00
Total Weekly Learner Workload					9.00
Total Weekly Contact Hours					5.00

Workload: Part Time					
Workload Type	Contact Type	Workload Description	Frequency	Average Weekly Learner Workload	Hours
Lecture	Contact	Delivery of content and materials underpinning learning outcomes	Every Week	3.00	3
Lab	Contact	Practical skills development	Every Week	2.00	2
Independent & Directed Learning (Non-contact)	Non Contact	Review of course material, completion of laboratory reports and exam practice	Every Week	2.00	2
Total Hours					7.00
Total Weekly Learner Workload					7.00
Total Weekly Contact Hours					5.00

## Module Resources

<b>Recommended Book Resources</b>
Brown, LeMay, Bursten, Murphy, Woodward, Stoltzfus. (2017), Chemistry: The Central Science in SI Units, 14th.. Pearson, [ISBN: 9781292221229].
<b>Supplementary Book Resources</b>
McMurray, Fay, Robinson. (2015), Chemistry, 7th. Pearson, [ISBN: 9781292092751].
<i>This module does not have any article/paper resources</i>
<b>Other Resources</b>
Website, Khan Academy. (2017), Khan Academy Chemistry, <a href="https://www.khanacademy.org/science/chemistry">https://www.khanacademy.org/science/chemistry</a> Website, Chemistry LibreTexts. (2016), Chemistry LibreTexts, <a href="https://chem.libretexts.org/">https://chem.libretexts.org/</a>

## Module Delivered in

Programme Code	Programme	Semester	Delivery
CR_SCHQA_8	<a href="#">Bachelor of Science (Honours) in Analytical Chemistry with Quality Assurance</a>	-1	Mandatory
CR_SESST_8	<a href="#">Bachelor of Science (Honours) in Environmental Science and Sustainable Technology</a>	-1	Mandatory
CR_SINEN_8	<a href="#">Bachelor of Science (Honours) in Instrument Engineering</a>	-1	Mandatory
CR_SCHEM_7	<a href="#">Bachelor of Science in Analytical and Pharmaceutical Chemistry</a>	-1	Mandatory
CR_SPHYS_7	<a href="#">Bachelor of Science in Applied Physics and Instrumentation</a>	-1	Mandatory
CR_ESBMO_6	<a href="#">Certificate in the Science of Biotechnological Manufacturing Operations</a>	-1	Mandatory
CR_SPHYS_6	<a href="#">Higher Certificate in Science in Applied Physics and Instrumentation</a>	-1	Mandatory
CR_SCHEM_6	<a href="#">Higher Certificate in Science in Chemistry</a>	-1	Mandatory
CR_SGMPR_6	<a href="#">Higher Certificate in Science in Good Manufacturing Practice and Technology</a>	-1	Mandatory
CR_SOMNI_8	<a href="#">Physical Sciences (Common Entry)</a>	-1	Mandatory
CR_SOMNI_7	<a href="#">Physical Sciences (Common Entry)</a>	-1	Mandatory